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54 Process for the preparation of iodoalkynyl carbamates.

57 A process for manufacture of iodoalkynyl carbamates
by reacting an alkynol with an isocyanate to give an alkynyl
carbamate, followed by iodination of the latter. The products
are fungicides, especially for paint films.

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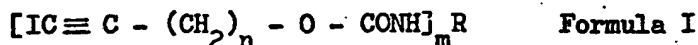
IMPERIAL CHEMICAL INDUSTRIES LIMITED

TITLE

PROCESS FOR THE PREPARATION OF IODOALKYNYL CARBAMATES

This invention relates to a chemical process and more particularly to a process for the manufacture of monoiodoalkynyl carbamates.

U.S. Patent Specification No. 3923870 discloses
5 compounds having the formula:



wherein R is a substituted or unsubstituted alkyl, aryl or alkylaryl group having from 1 to 20 carbon atoms and having from one to three linkages corresponding to m , and
10 m and n are whole number integers between 1 and 3 and may be the same or different. These compounds have fungicidal activity, and are particularly useful as paint film fungicides. The only method disclosed for their preparation is iodination of an alkynol of formula $\text{HC} \equiv \text{C} \cdot (\text{CH}_2)_n - \text{OH}$
15 to give $\text{IC} \equiv \text{C} \cdot (\text{CH}_2)_n - \text{OH}$, followed by reaction of the latter with an isocyanate of formula $\text{R}(\text{NCO})_m$, wherein R, m and n have the previously defined meanings.

This process has the disadvantages that many of the intermediate iodoalkynols of formula $\text{IC} \equiv \text{C} \cdot (\text{CH}_2)_n - \text{OH}$
20 are highly vesicant and also tend to be mildly explosive. These properties make manufacture by this route both unpleasant and potentially dangerous.

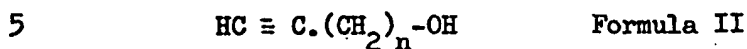
An alternative route has now been found which avoids these disadvantages.

25 According to the present invention there is provided a process for the manufacture of compounds having the formula:



wherein R is a substituted or unsubstituted alkyl, cycloalkyl,

aryl, arylalkyl or alkylaryl group having from 1 to 20 carbon atoms and having m linkages and \underline{m} and \underline{n} are independent integers from 1 to 3, which comprises reacting an alkynol having the formula:



with an isocyanate having the formula $\text{R}(\text{NCO})_m$ to give a carbamate having the formula:



followed by iodination of the said carbamate.

10 The alkynols of Formula II in which \underline{n} is 1, 2 or 3 are known compounds. Prop-1-yn-3-yl alcohol and but-1-yn-4-ol are mentioned in Volume 1 of Beilstein's Handbuch der Organischen Chemie at pages 454 and 455 respectively. Pent-1-yn-5-ol is disclosed in Chemical
15 Abstracts, Vol.26, page 2166 (1932).

As examples of suitable isocyanates of the formula $\text{R}(\text{NCO})_m$ there may be mentioned alkyl isocyanates, such as octadecyl, dodecyl, octyl, hexyl and especially lower alkyl isocyanates such as butyl, propyl, ethyl and
20 methyl isocyanates and the various structural isomers thereof, cycloalkyl isocyanates, such as cyclohexyl isocyanate and monocyclic aryl isocyanates such as phenyl, 4-chlorophenyl and 3,4-dichlorophenyl isocyanates in which m is 1 and alkylene diisocyanates such as
25 hexamethylene diisocyanate, arylene diisocyanates such as methylene bisphenyl diisocyanate and the separate or mixed isomers of tolylene diisocyanate and alicyclic diisocyanates such as isophorone diisocyanate in which m is 2.

As examples of the optional substituents that may be present on the above mentioned isocyanates there may be mentioned halogens, such as chloro, bromo and fluoro and lower, i.e. C_1 to C_4 , alkyl groups.

5 The first stage of this process, reaction of an alkynol of Formula II with an isocyanate $R.(NCO)_m$, is carried out by methods which are well known in the urethane art. Thus, the isocyanate may be added to the alkynol at a rate such that the reaction temperature is readily controllable,
10 if desired with the aid of external cooling. The reaction may be carried out in the absence of a solvent, or a solvent which is inert to isocyanate groups may be used. Examples of suitable solvents are esters such as ethyl acetate, butyl acetate and 2-ethoxyethyl acetate, ketones such as methyl
15 ethyl ketone, methyl isobutyl ketone, cyclohexanone and 4-methoxy-4-methylpentan-2-one, hydrocarbons such as toluene and xylene, halogenated hydrocarbons such as trichloroethylene, methylene chloride, 1,1,1-trichloroethane and tetrachloroethane, and polar aprotic solvents such as
20 dimethylformamide and dimethylacetamide.

There may also be used a catalyst of the kind which accelerates reaction between an isocyanate group and a hydroxyl group. As examples of such catalysts there may be mentioned organic and inorganic basic compounds, and soluble compounds
25 of metals, for example, of transition metals, such as iron and manganese acetylacetonate, and of tin and antimony, for example, dibutyl tin dilaurate and stannous octoate; compounds of lead such as lead acetate, basic lead acetate and lead 2-ethylhexoate. As basic organic catalysts

tertiary amines are suitable, particularly triethylamine, 4-dimethylaminopyridine, triethylenediamine, dimethylbenzylamine, and dimethylcyclohexylamine. Particularly advantageous are mixtures of metallic and amine catalysts. The amount of catalyst used will depend upon its activity, the temperature used and the rate of reaction desired, but in general an amount from 0.02 to 1.0% by weight, based on the weight of alkynol, will be appropriate. The reaction is conveniently initiated at normal room temperature, e.g. 15 to 25°C, or at a moderately elevated temperature up to 40°C, the temperature preferably being raised to 50-60°C or somewhat higher, e.g. up to 100°C, to promote completion of the reaction. The alkynylcarbamate is isolated from the reaction mixture by conventional means.

The second stage of the process, iodination of the alkynyl carbamate may be carried out by reaction of the latter in aqueous medium with an iodinating agent, such as a mixture of sodium hypochlorite and an alkali metal iodide, or a mixture of sodium hypochlorite, an alkali metal hydroxide and iodine. A slight molar excess of the iodinating agent is preferably employed to favour complete conversion of the alkynylcarbamate into its iodo derivative.

In order to promote contact between the sparingly soluble alkynyl carbamate and the iodinating agent, the reaction is conveniently performed in the presence of an auxiliary solvent, preferably organic, for the alkynyl carbamate which is also miscible with water. As examples of such solvents there may be mentioned alkanols such as methanol and ethanol. The alkynyl carbamate may alternatively be dispersed or emulsified in an aqueous medium preferably with the aid of a surfactant so as to maintain it in a finely divided state throughout the reaction period and thereby to promote contact with the iodinating agent. If the alkynyl carbamate is suitably dispersed or emulsified it is not normally necessary to employ an auxiliary solvent.

Classes of substances suitable for use as a surfactant during iodination are protective colloids, such as hydrolysed or partially hydrolysed polyvinyl acetates, phase transfer catalysts such as cetyl trimethyl ammonium bromide, dispersing agents such as the sodium salt of a condensate of formaldehyde and naphthalene sulphonic acid, and polyglycerol ricinoleate, wetting agents such as the sodium salt of di-isopropyl naphthalene sulphonic acid and emulsifying agents such as ethoxylated nonyl phenols.

10 The effect of the surfactant may be improved by the use of a high speed agitator in the reaction vessel and preferably one which provides some shear action, such as a Silverson or a Turrax mixer. (Silverson and Turrax are trade names).

15 The temperature during the second stage of the process is preferably maintained at or below 10°C.

The product, i.e. the compound of Formula I, is conveniently isolated by neutralising the reaction mixture with an acid, such as hydrochloric acid, followed by extraction of the iodoalkynyl carbamate from the reaction mixture with a water-immiscible solvent, such as toluene. The product may then be separated from the solvent and used as such or the product may be used as a solution in the solvent as such or after the addition of an emulsifying agent to make the solution self-emulsifiable. Alternatively the product may be dispersed in water or some other medium in which it is substantially insoluble with the aid of appropriate surfactants.

25 30 The invention is illustrated by the following Examples in which parts and percentages are by weight.

Example 1(a) Preparation of prop-1-yn-3-yl N-n-butylcarbamate

Prop-1-yn-3-yl alcohol (56 parts), toluene (250 parts), triethylamine (0.257 part) and dibutyl tin dilaurate (0.257 part) are charged to a stirred reaction vessel and n-butyl isocyanate (99 parts) is added slowly and continuously during 1 hour. The reaction temperature is controlled between 22° and 37°C by intermittent external cooling. The reaction mixture is then heated to 50°C for six hours, and cooled.

The toluene solution is extracted with 10% sodium carbonate solution (50 parts) and then washed alkali-free with water. The toluene is then evaporated under vacuum at 30°C to leave a quantitative yield of prop-1-yn-3-yl N-n-butylcarbamate as a pale yellow liquid. The infra-red spectrum of this product shows bands at 3320cm^{-1} , 2135cm^{-1} and 1710cm^{-1} .

(b) Preparation of 1-iodoprop-1-yn-3-yl N-n-butylcarbamate

Ethanol (60 parts), water (25 parts) and prop-1-yn-3-yl N-n-butylcarbamate (7.75 parts) are charged to a stirred reaction vessel, followed by sodium hydroxide (3 parts) dissolved in the minimum amount of water. The solution is cooled to $\leq 10^\circ\text{C}$ and iodine (7 parts), followed by sodium hypochlorite (20.5 parts, as a 6% aqueous solution) is added, maintaining the temperature at $\leq 10^\circ\text{C}$. The solution is stirred at this temperature and hydrochloric acid is added to adjust the pH to 7. After allowing the solution to warm to room temperature it is extracted with several portions of toluene. The toluene extracts are combined and dried over anhydrous magnesium sulphate, and the toluene is then removed by evaporation under vacuum to yield 1-iodoprop-1-yn-3-yl N-n-butylcarbamate (8.56 parts), m.p. 65-67°C. The N-n-infra-red spectrum of this product shows bands at 3340cm^{-1} , 2200cm^{-1} and 1705cm^{-1} .

Example 2

Ethanol (60 parts), water (25 parts) and prop-1-yn-3-yl N-n-butylcarbamate (7.75 parts) are charged to a stirred reaction vessel, followed by sodium iodide (8.25 parts). The resulting solution is cooled to $\leq 10^{\circ}\text{C}$, and sodium hydroxide (2 parts) and sodium hypochlorite (4.7 parts, as a 6% aqueous solution) are added at such a rate as to maintain this temperature. After the reaction mixture has been stirred at $\leq 10^{\circ}\text{C}$ for 5 hours sodium bisulphite (5.57 parts as a 40% aqueous solution) is added and the mixture is then neutralised with hydrochloric acid. The solution is extracted with portions of toluene, the toluene extracts are combined and dried over anhydrous magnesium sulphate, and the toluene is removed by evaporation under reduced pressure to yield 1-iodoprop-1-yn-3-yl N-n-butylcarbamate (8.7 parts).

Example 3

Prop-1-yn-3-yl N-n-butylcarbamate (7.75 parts) and methanol (80 parts) are charged to a stirred reaction vessel and sodium hydroxide (2.89 parts) dissolved in the minimum amount of water is added. The solution is cooled to $\leq 10^{\circ}\text{C}$ and iodine (13.33 parts) is added portionwise during 20 minutes, maintaining the temperatures at $\leq 10^{\circ}\text{C}$. The solution is maintained at this temperature for a further 40 minutes before being allowed to rise to room temperature. After 5 hours the solution is extracted with several portions of toluene, the toluene extracts are combined and dried over anhydrous magnesium sulphate, and the toluene is removed by evaporation under reduced pressure to yield 1-iodoprop-1-yn-3-yl N-n-butylcarbamate (11.2 parts).

Example 4

Prop-1-yn-3-yl N-n-butylcarbamate (15.5 parts) is stirred with sodium hydroxide (0.8 part), potassium iodide (15.5 parts) and a partially hydrolysed polyvinyl acetate sold under the trade name of Gohsenol GL-03 (0.5 part). Sodium hypochloride (8.94 parts) is added as an 11.5% aqueous solution during 75 minutes with control of the temperature to 1-3°C. The white solid dispersion is stirred for a further 30 minutes then a sample examined by the infra-red spectrometer. The absorption at 2110cm^{-1} characteristic of a $\text{HC} \equiv \text{C}$ - residue is absent. That at 2200cm^{-1} characteristic of an $\text{I} - \text{C} \equiv \text{C}$ - unit is present.

Example 5

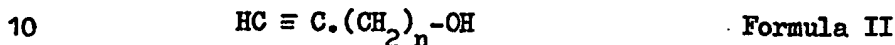
Prop-1-yn-3-yl N-n-butylcarbamate (24.8 parts) is stirred with water (70 parts), polyglycerol ricinoleate (Dispersol OG) (2.6 parts) and iodine (21.4 parts). Sodium hydroxide (25 parts) is added as a 32% aqueous solution during 30 minutes with control of the temperature at $\leq 10^\circ\text{C}$ followed by sodium hypochlorite (62 parts) as a 12% aqueous solution, again at $\leq 10^\circ\text{C}$. The dispersion is stirred for a further 30 minutes then a sample examined by an infra-red spectrometer. The absorption at 2110cm^{-1} characteristic of a $\text{CH} \equiv \text{CH}$ residue is absent. That at 2200cm^{-1} characteristic of an $\text{I} - \text{C} \equiv \text{C}$ unit is present.

CLAIMS

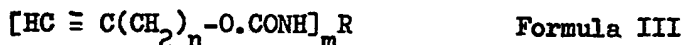
1. A process for the preparation of a compound of the formula:



- 5 wherein R is a substituted or unsubstituted alkyl, cycloalkyl, aryl, aralkyl or alkylaryl group having from 1 to 20 carbon atoms and having m linkages and m and n are independent integers from 1 to 3, which comprises reacting an alkynol having the formula:



with an isocyanate having the formula $R.(NCO)_m$ to give a carbamate having the formula:



followed by iodination of the said carbamate.

- 15 2. A process according to Claim 1 wherein R is selected from the group consisting of C_1 to C_4 alkyl, 4-chlorophenyl, 3,4-dichlorophenyl, phenyl, hexamethylene, isophorone-di-yl, tolylene and methylene bisphenyl.
3. A process according to Claim 1 or Claim 2 wherein
20 the iodination of the carbamate of Formula III is performed in an aqueous medium in the presence of an auxiliary water-miscible organic solvent.
4. A process according to Claim 1 or Claim 2 wherein
25 the carbamate of Formula III is iodinated in the form of a dispersion or emulsion in an aqueous medium.
5. A process according to Claim 4 wherein the carbamate of Formula III is dispersed or emulsified with the aid of a surfactant.

6. A process according to Claim 5 wherein the surfactant is selected from the group comprising a protective colloid, a phase transfer catalyst, a dispersing agent, a wetting agent and an emulsifying agent.
- 5 7. A process according to Claim 5 or Claim 6 wherein the surfactant is selected from the group comprising hydrolysed or partially hydrolysed polyvinyl acetate, cetyl trimethyl-ammonium bromide, the sodium salt of a condensate of formaldehyde and naphthalene sulphonic acid, polyglycerol
- 10 ricinoleate, the sodium salt of di-isopropyl naphthalene sulphonic acid and ethoxylated nonylphenol.
8. A process according to any one of Claims 5 to 7 which is performed in a vessel containing a high speed agitator.

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